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[3] F1. Crude oil contains many hydrocarbons, including hexane,  $C_6H_{14}$ . (a) Outline how crude oil was formed..... [3] (b) The equation for the complete combustion of hexane is shown below.  $C_6H_{14}(g) + 9O_2(g) \rightarrow 6CO_2(g) + 7H_2O(g)$  Determine a value for the enthalpy of combustion of hexane using the following enthalpy of formation data ...

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