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Vocabulary from Chapter 2 of the Modern Chemistry textbook. (20 words in total) Terms in this set (20) scientific method, a series of steps followed to solve problems, including collecting data, formulating a hypothesis, and stating conclusions. system.

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the explanation for some phenomenon that is based on observation, experimentation, and reasoning; that is supported by a large quantity of evidence; and that does not conflict with any existing experimental results or observations. quantity, something that has magnitude, size, or amount. SI.

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Modern Chemistry Chapter 2. STUDY. PLAY. scientific method, a logical approach to solving problems by observing and collecting data, formulating hypotheses, testing hypotheses, and formulating theories that are supported by data. hypothesis.

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chapter 2 modern chemistry Flashcards and Study Sets | Quizlet A B; atom: the basic unit of matter: nucleus: the center of the atom: electron: a negatively charged particle: element: a pure substance that consists entirely of one type of atom

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Modern Chemistry Chapter 2 Vocabulary

a substance formed by the chemical combination of two or more elements in definite proportions. ionic bond, formed when one or more electrons are transferred from one atom to another. ion, positively and negatively charged atoms. covalent bond, forms when electrons are shared between atoms. molecule.

Quia - Chapter 2: The Chemistry of Life Vocabulary Review

a homogeneous mixture of two or more substances uniformly dispersed throughout a single phase. Phase, any part of a sample with uniform composition and properties (oil and water would consist of two phases, the oil is one, and the water is another) Filtration, the process that separates a solid from the liquid in a heterogeneous mixture.

Study Chapter 2 Vocabulary: Prentice Hall Chemistry ...

Terms in this set (32) chemistry, the scientific study of the composition, structure, and properties of matter and the changes that matter undergoes. chemical, any substance that has a defined composition. mass, a measure of the amount of matter in an object. matter, anything that has mass and takes up space.

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2 (g) Æ 2NaCl(s). a. How many grams of sodium are required to produce 378.5 g of salt? b. How many grams of Cl 2 are needed? 2. Calcium reacts with oxygen in air to produce calcium oxide in the reaction $2\text{Ca}(s) + \text{O}_2(g) \text{Æ} 2\text{CaO}(s)$. How many grams of oxygen gas must react to yield 20.55 g of calcium oxide? 3. For the reaction $2\text{LiOH} + \text{CO}_2 \text{Æ} \text{Li}_2\text{CO}_3 + \text{H}_2\text{O}$

HOMEWORK 9-1

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CH 1 Reading Assignment Modern Chemistry, CH 1 Vocabulary-New, CH 1 Mixed Questions, CH 1 Matter & Energy Vocabulary-New, CH 1 Classifying Matter-New, CH 1 Classification of Matter ... Notes: Chapter 2 CH 2 Measurement & Calculations Honors Chapter 2. Reference: Sig Fig Rules, Metric Units Conversion, Review Activities: ...