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2 (g) + NO₂ (g) In one such reaction 0,74 g of O₃ reacts with 0,67 g NO. 9.1 Calculate the number of moles of O₃ and of NO present at the start of the reaction. (5) 9.2 Identify the limiting reagent in the reaction and justify your answer. (2) 9.3 Calculate the mass of NO₂ produced from the

reaction. (4) [11] QUESTION 10

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